

# Octadeca-8,11-dienoic Acid Methylester, a New Fatty Acid Metabolite from *Fistulina hepatica*

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**Abstract:** A new fatty acid metabolite, octadeca-8,11-dienoic acid methylester (**1**) was isolated together with the known ergosterol (**2**) from *Fistulina hepatica*. The edible mushroom *Fistulina hepatica* is a relatively rarely occurring representative of the wood-rotting basidiomycetes for which no characteristic metabolites have been reported so far. The compounds **1** and **2** were purified by solvent extraction, silica gel column chromatography and preparative HPLC consecutively. The structures of the both compounds were unequivocally determined on the basis of NMR spectroscopic methods including 1D (<sup>1</sup>H, <sup>13</sup>C and DEPT) and 2D (COSY, HSQC and HMBC) and by mass spectrometry data (ESI, EI and HREI-MS). The new fatty acid methylester **1** displays no antimicrobial properties against *Staphylococcus aureus*, *Bacillus cereus*, *Bacillus subtilis*, *Pseudomonas aeruginosa* and *Candida albicans*, but very low activity against *Escherichia coli* and *Proteus vulgaris*.

**Keywords:** *Fistulina hepatica*, octadeca-8,11-dienoic acid methylester, ergosterol.

## INTRODUCTION

Large quantity of natural products with different structure types have not only been focused on the study of functional molecules for the organism in the fields of chemical defense, chemical regulation and signal transduction, but also played a vital role in the discovery of new pharmacy. Germany, with its climatic conditions and flora diversity, is one of the European regions with higher wild edible mushrooms diversity, some of them with great gastronomic relevance. Studies conducted on mushrooms proved their antioxidant [1], antitumor [2], and antimicrobial properties, as well as their interesting contents in nutraceuticals [3]. Furthermore, mushrooms are becoming important in our diet for their nutritional and organoleptic characteristics [4]. Many authors have been interested in the bioactive properties and chemical profile of wild and commercial mushrooms; regarding chemical characterization, special attention have been dedicated to the determination of proteins, carbohydrates, individual sugars, fatty acids, phenolic compounds, carotenoids, ascorbic acid and tocopherols [5-11]. Other authors also analyzed ergosterol, vitamin D<sub>2</sub>, nucleosides and nucleobases in mushrooms [12-15]. A comparative study was developed on the total fatty acids composition of twelve wild edible mushroom species (*Suillus bellini*, *Suillus*

*luteus*, *Suillus granulatus*, *Hygrophorus agathosmus*, *Amanita rubescens*, *Russula cyanoxantha*, *Boletus edulis*, *Tricholoma equestre*, *Fistulina hepatica*, *Cantharellus cibarius*, *Amanita caesarea* and *Hydnum rufescens*). By GC-MS analysis was established that polyunsaturated and monounsaturated fatty acids, valuable healthy compounds for humans, predominated over saturated fatty acids for all the studied mushroom species. *R. cyanoxantha* presented the highest fatty acids amounts, while *B. edulis* was the poorest species [16].

In this paper, we describe the isolation, structure elucidation and the antimicrobial properties of the related new octadeca-8, 11-dienoic acid methylester (**1**) and the known ergosterol (**2**), using high resolution EI-MS, ESI-MS and extensive NMR spectroscopic analysis.

## EXPERIMENTAL

### General Experimental Procedures

Electrospray mass spectra (ESI-MS) were recorded on a LCQ Finnigan Mass Spectrometer. ESI-HR mass spectra were recorded on a APEX IV, 7T, FT-ICR MS Bruker Daltonik. Electron Impact Mass-spectra (EI-HRMS) were recorded on a double focusing sector field mass spectrometer AMD-402 (AMD Inetra, Harpstedt, Germany). <sup>1</sup>H (300 and 600 MHz) and <sup>13</sup>C (75.5 and 125.7 MHz) NMR spectra were measured on a Bruker AMX 300 and on a Varian Inova 600 (599.740 MHz) spectrometer in CDCl<sub>3</sub> and CD<sub>3</sub>OD. The chemical shifts

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are expressed in  $\delta$  values with TMS as an internal standard.  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC and HMBC were obtained by conventional methods. UV-Vis spectroscopy was recorded on a Specord 2000 instrument (Analytik Jena, Germany). IR spectra were recorded on a Beckman DU 601 and Shimadzu IR scanning spectrophotometers. Gas-chromatography was performed with a Shimadzu instrument, coupled to a FID detector.

### Producing Organism

The edible mushroom *Fistulina hepatica* grows widespread in the Northern hemisphere on oak trees and attracts attention due to its excellent taste and high nutrient value. A single fruit-body of *Fistulina hepatica* was collected in the forestal region of Seitenbrueck near Jena, Germany.

### Isolation of a New Fatty Acid, Octadeca-8,11-dienoic Acid Methyleneester (1) and Ergosterol (2) from the Basidiomycete *Fistulina hepatica*

The biomass of *Fistulina hepatica* (550 g) was extracted threetimes with 1000 ml methanol. The combined methanol extracts were concentrated *in vacuo*, and the remaining watery phase was extracted with 3 x 500 ml ethyl acetate. After evaporation of the solvent the crude product (1.2 g) was dissolved in chloroform and chromatographed on a silica gel 60 (70-230 mesh) column (5 cm x 300 cm). The metabolites were eluted from the column by stepwise use of *n*-hexane (800 ml), chloroform (1000 ml), chloroform-methanol (9:1/750 ml; 7:3/1000 ml; 5:5/500 ml; 2:8/500 ml; 1:10/300 ml) and methanol (800 ml). The compound **2** was eluted with chloroform-methanol (7:3/1000 ml) and the fraction was evaporated to dryness under reduced pressure, dissolved in methanol and crystallised. Compound **1** was eluted with chloroform-methanol (1:10/300 ml) and detected by its diagnostical reddish staining by 3% vanillin/methanol/ $\text{H}_2\text{SO}_4$  as reagent. The final separation and purification of **1** by preparative HPLC on Spherisorb ODS2 (column 12.5 mm x 250 mm), using an acetonitrile-water (83:17, v/v) and monitoring at 210 nm (flow rate, 10.0 ml min<sup>-1</sup>) delivered the pure octadeca-8,11-dienoic acid methyleneester **1** (8.0 mg) as waxy colorless solid.

### Analytical HPLC

Analytical HPLC analysis of pure octadeca-8,11-dienoic acid methyleneester **1** was carried out on a Spherisorb ODS2 (column 12.5 mm x 250 mm). Mobile

phase: acetonitrile-water (83:17, v/v). Flow rate: 1.0 ml/min. Detection: 210 nm (diode array detector). Compound **1** was eluted with a retention time ( $t_R$ ) of 2.10 minutes.

### Thin-Layer Chromatography (TLC)

TLC was carried out on silica gel plates (Merck 60, F<sub>254</sub>) with *n*-butanol-acetic acid-water [5:1:4 v/v, (upper phase)] of compound **1** and chloroform-methanol (9:1 v/v) for compound **2**. The chromatographic spots were visualized by spraying with a 0.5% solution of vanillin in methanol/sulfuric acid/acetic acid and ammonium molybdate, and heating at 120 °C for 3-5 minutes. In this systems the compounds **1** and **2** possessed an  $R_f$  values of 0.45 for **1** and 0.67 for **2**, and gave a blue-violet, resp. a blue color.

### Antimicrobial Activity of Compound 1

Antimicrobial activity was determined by agar diffusion test according to European Pharmacopoeia, 1997 [17]. Test organisms were suspended in the melted nutrient agar (Serva) and poured into petri dishes. Holes of 8 mm diameter were cut in the agar and filled with 100  $\mu\text{l}$  of a 100 mg l<sup>-1</sup> solution of the compound. Inhibition zones were read after incubation for 18 h at 37 °C for bacterial strains and 30 °C for yeasts.

## RESULTS AND DISCUSSION

Compound **1** was obtained as a waxy colorless solid, which was soluble in methanol, ethanol, dimethylsulfoxide, *N,N*-dimethylformamide and ethyl acetate, but insoluble in water, ether and *n*-hexane. It gave a blue-violet and blue color positive reactions with 3% vanillin-sulfuric acid and ammonium heptamolybdate [(NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>]. The physico-chemical properties of octadeca-8,11-dienoic acid methyleneester (**1**) are summarized in Table 1. The presence of carboxyl acid ester group was attested by  $\nu_{\text{max}}$  1734 cm<sup>-1</sup> in the IR spectrum. The chemical constitution and stereochemistry of **1** were settled on the basis of electrospray (ESI-MS), high-resolution electron impact mass spectrometry (EI-HRMS) and one- and two dimensional NMR measurements ( $^1\text{H}$ ,  $^{13}\text{C}$ , DEPT,  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC and HMBC).

The (+)-ESI mass spectrum of **1** displayed *pseudo*-molecular ion at  $m/z$  317.3 (M+Na)<sup>+</sup>. The molecular weight (294 g/mol) and the chemical formula C<sub>19</sub>H<sub>34</sub>O<sub>2</sub> were readily determined by high-resolution electron impact mass spectrometry (EI-HRMS,  $m/z$  294.25460).

Table 1: Physico-Chemical Properties of Octadeca-8,11-dienoic Acid Methylene Ester (1)

Appearance	waxy colorless solid
Molecular formula	C <sub>19</sub> H <sub>34</sub> O <sub>2</sub>
Molecular weight	294 g/mol
EI-HRMS (m/z):	calcd.: 294.25321
	found: 294.25460
UV-Vis λ <sub>max</sub> CH <sub>3</sub> OH (nm)	< 230
IRν <sub>max</sub> (KBr, cm <sup>-1</sup> )	524, 624, 651, 717, 979, 1090, 1182, 1249,
	1380, 1404, 1457, 1579, 1734, 2913, 2925,
	3425
R <sub>f</sub> on TLC	0.45
[silica gel sheets Merck, Mobile phase:n-butanol-acetic acid-water, 5:1:4 v/v, (upper phase)]	
T <sub>R</sub> on analytical HPLC (min)	2.10
[Spherisorb ODS2 (column 12.5 mm × 250 mm) Mobile phase: acetonitrile-water (83:17, v/v)]	

This formula indicated 3 double bond equivalents. EI-MS of **1** exhibited a strong ion peak at  $m/z$  294.2, together with fragment peaks at  $m/z$  270.2 (C<sub>17</sub>H<sub>34</sub>O<sub>2</sub>), 239.2 (C<sub>16</sub>H<sub>31</sub>O), 227.1 (C<sub>14</sub>H<sub>27</sub>O<sub>2</sub>) and 143.1 (C<sub>8</sub>H<sub>15</sub>O<sub>2</sub>).

In the <sup>1</sup>H NMR spectrum of compound **1**, four olefinic protons appeared at δ<sub>H</sub> 5.34 (H-8), 5.33 (H-9), 5.34 (H-11) and 5.31 (H-12). The remaining signals of the proton spectrum included a large signals at δ<sub>H</sub> 1.27÷2.35 due to a number of overlapping methylene signals as well, as a methyl protons at δ<sub>H</sub> 0.95 and the protons of the methoxy group at δ<sub>H</sub> 3.64. The all-*cis* configuration at the two double bonds (C-8, C-9, C-11, C-12) was proved by <sup>3</sup>J<sub>H-8, H-9</sub> = 6.0 Hz and <sup>3</sup>J<sub>H-11, H-12</sub> = 6.0 Hz, typical for a *cis*-olefin. Catalytic hydrogenation of **1** by Pt/H<sub>2</sub> yielded stearic acid methylester as was shown by GC measurement referring to authentic material.

The <sup>13</sup>C and DEPT NMR spectral data of **1** presented in Table 2 suggested the presence of one carboxylic acid ester group (δ<sub>C</sub> 180.34), four sp<sup>2</sup> carbons (δ<sub>C</sub> 129.06, 129.86, 130.86 and 130.93), one methyl group (δ<sub>C</sub> 14.20), one methoxyl (δ<sub>C</sub> 51.90) and 12 methylene groups). The chemical shifts of all proton and carbon atoms of **1** are summarized in Table 2.

For the structural assignment of compound **1**, the <sup>1</sup>H-<sup>1</sup>H COSY and <sup>13</sup>C-<sup>1</sup>H long-range heteronuclear coupled NMR spectra (HMBC) were of pivotal importance.

According to the <sup>1</sup>H-<sup>1</sup>H COSY spectra, the methyl group (H<sub>3</sub>-18) at δ<sub>H</sub> 0.95 was linked to the methylene

group (H<sub>2</sub>-17) at δ<sub>H</sub> 1.30 adjacent to the methylene group at δ<sub>H</sub> 1.60 (H<sub>2</sub>-16). The methylene group (H<sub>2</sub>-15) at δ<sub>H</sub> 2.35 was linked to the methylene group (H<sub>2</sub>-14) at δ<sub>H</sub> 1.36 adjacent to the methylene group (H<sub>2</sub>-13) group at δ<sub>H</sub> 2.05. The COSY experiment demonstrated also couplings from the olefinic proton of H-11 at δ<sub>H</sub> 5.34 to the protons of the methylene group (H<sub>2</sub>-10) at δ<sub>H</sub> 2.75. Similarly a coupling was established from the olefinic proton of H-8 at δ<sub>H</sub> 5.34 to the protons of the methylene group (H<sub>2</sub>-7) at δ<sub>H</sub> 2.05.

A heteronuclear multiple bond correlation (HMBC) experiment showed long-range couplings from the protons of H<sub>3</sub>-19 at δ<sub>H</sub> 3.64 and H<sub>2</sub>-2 at δ<sub>H</sub> 1.88 to the carbonyl carbon C-1 at δ<sub>C</sub> 180.3. Further correlations are demonstrated in Table 2 and Figure 2.

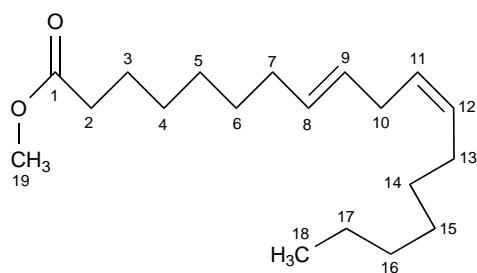
Thus octadeca-8,11-dienoic acid (**1**) appears as a new olefinic fatty acid occurring as methylester (Figure 1) in *Fistulina hepatica*. **1** enlarges the spectrum of the know unsaturated fatty acids from basidiomycetes. The NMR data enabled doubtlessly to distinguish between **1** and related fatty acids such as octadeca-9,12-dienoic acid (linolenic acid) as a frequently occurring plant product. Obviously in *Fistulina hepatica* there are desaturating enzymes introducing two isolated (*Z*)-double bonds into a C18-aliphatic acid. The new fatty acid methylester **1** displays no antimicrobial properties against *Staphylococcus aureus*, *Bacillus cereus*, *Bacillus subtilis*, *Pseudomonas aeruginosa* and *Candida albicans*, but very low activity (≥1000 μg/ml) against *Escherichia coli* and *Proteus vulgaris*. But due to the activated C-10 methylene group **1** should readily be oxidized in presence of singlet oxygen and thus could

**Table 2:**  $^1\text{H}$  and  $^{13}\text{C}$  NMR Chemical Shifts of Octadeca-8,11-dienoic Acid Methyl ester (1) in  $\text{CD}_3\text{OD}$  (600 MHz and 125.7 MHz, TMS as Internal Standard, Chemical Shifts in  $\delta$  Values)

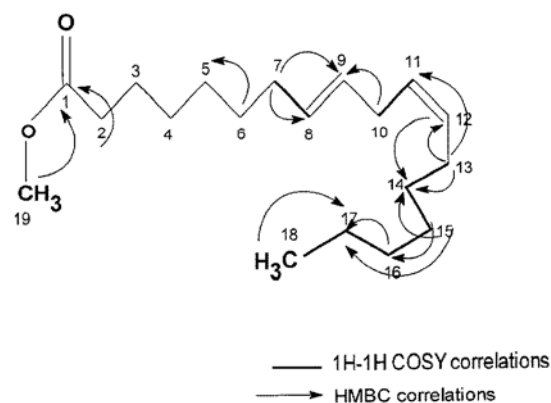
Assignment	$\delta_{\text{H}}$ , J (Hz)	HMBC (H $\rightarrow$ C)	$\delta_{\text{C}}$	Groups
1	-		180.3	C=O
2	1.88 (2H, t,br)	C-1	23.9	CH <sub>2</sub>
3	1.27 (2H, m)		30.9	CH <sub>2</sub>
4	1.27 (2H, m)		30.9	CH <sub>2</sub>
5	1.27 (2H, m)		30.9	CH <sub>2</sub>
6	1.36 (2H, m)	C-5	30.2	CH <sub>2</sub>
7	2.05 (2H, m)	C-8, C-9	28.0	CH <sub>2</sub>
8	5.34 (1H, dd, 6.0, 6.0 Hz)		130.9	HC=
9	5.33 (1H, dd, 6.0, 4.0 Hz)		129.1	HC=
10	2.75 (2H, d,d,d, 6.0, 4.0, 1.0)	C-9	27.0	CH <sub>2</sub>
11	5.34 (1H, dd, 6.0, 6.0 Hz)		130.9	HC=
12	5.31 (1H, dd, 6.0, 4.0 Hz)		129.0	HC=
13	2.05 (2H, d, 4.0 Hz)	C-11, C-12, C-14	28.0	CH <sub>2</sub>
14	1.36 (2H, m)		30.2	CH <sub>2</sub>
15	2.35 (2H, m)	C-14, C-16, C-17	35.0	CH <sub>2</sub>
16	1.60 (2H, m)	C-17	26.0	CH <sub>2</sub>
17	1.30 (2H, m)		23.4	CH <sub>2</sub>
18	0.95 (3H, dd, 7.0 Hz)	C-17	14.2	CH <sub>3</sub>
19	3.64 (3H, s)	C-1	51.9	H <sub>3</sub> C-O

<sup>†</sup>Coupling constants in Hz; s: singlet, d: doublet, t: triplet, m: multiplet, br: broad.

be considered as a powerful radical scavenger and valuable antioxidant food additive.

**Figure 1:** Structure of octadeca-8,11-dienoic acid methyl ester (1).

By Gas-chromatography of the fraction, eluted with methanol from the silica gel 60 (70-230 mesh) column from *Fistulina hepatica* was established also presence of the following fatty acids: dodecanoate (C12:0), 34.1%; hexadecanoate (C16:0), 21.9%; octadecanoate (C18:0), 9.2%; octadecenoate acid (C18:1, *cis* 9), 10.6% and octadecadienolic acid (C18:2, *cis* 9,12), 1.5%.

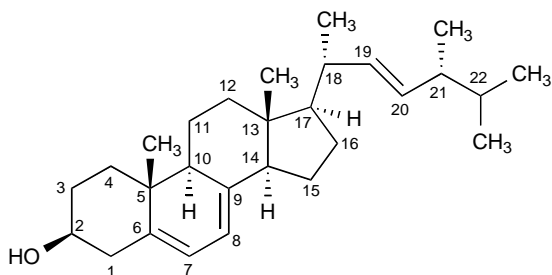
**Figure 2:** Instructive  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC correlations of octadeca-8,11-dienoic acid methyl ester (1).

Compound **2** was isolated as a colorless crystals, which were soluble in chloroform, ethyl acetate but insoluble in lower alcohols and water. The chemical constitution of **2** was settled on the basis of high-resolution electron impact mass spectrometry (EI-HRMS) and one- and two dimensional NMR measurements ( $^1\text{H}$ ,  $^{13}\text{C}$ , DEPT,  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC

**Table 3:**  $^{13}\text{C}$  NMR and DEPT Chemical Shifts of Ergosterol (2) in  $\text{CDCl}_3$  (125.7 MHz, TMS as Internal Standard, Chemical Shifts in  $\delta$  Values)

Assignment	$\delta_c$	Groups	Assignment	$\delta_c$	Groups
1	40.8	$\text{CH}_2$	13- $\text{CH}_3$	12.0	$\text{CH}_3$
2	70.5	$\text{CH-O}$	14	54.5	$\text{CH}$
3	31.9	$\text{CH}_2$	15	23.0	$\text{CH}_2$
4	38.4	$\text{CH}_2$	16	40.0	$\text{CH}_2$
5	37.0	Cq	17	55.7	$\text{CH}$
5- $\text{CH}_3$	16.3	$\text{CH}_3$	18	40.4	$\text{CH}$
6	141.3	Cq	18- $\text{CH}_3$	21.1	$\text{CH}_3$
7	119.6	$\text{HC=}$	19	132.0	$\text{HC=}$
8	116.3	$\text{HC=}$	20	135.5	$\text{HC=}$
9	139.8	Cq	21	42.8	$\text{CH}$
10	46.3	$\text{CH}$	21- $\text{CH}_3$	17.6	$\text{CH}_3$
11	21.1	$\text{CH}_2$	22	33.1	$\text{CH}$
12	28.3	$\text{CH}_2$	22- $\text{CH}_3$	19.9	$\text{CH}_3$
13	42.8	Cq	22- $\text{CH}_3$	19.6	$\text{CH}_3$

and HMBC). The molecular weight (396 g/mol) and the chemical formula  $\text{C}_{28}\text{H}_{44}\text{O}$  were readily determined by high-resolution electron impact mass spectrometry (EI- HRMS,  $m/z$  396.24360). This formula indicated 7 double bond equivalents. The chemical shifts of all carbon atoms of **2** are summarized in Table 3. On the basis of the above evidence the structure of **2** was established as known ergosterol [ergosta-5,7,22-trien-3 $\beta$ -ol] (Figure 3). It is a sterol found in *Fistulina hepatica*. Ergosterol does not occur in plant or animal cells. It is a component of yeast and fungal cell membranes, serving the same function cholesterol serves in animal cells. However, ergosterol is not produced by all fungi and the ergosterol concentrations are known to vary between the same species depending on the physiological state of the fungus. Ergosterol is an unsaturated hydrocarbon of the vitamin D group isolated from yeast, mushrooms, ergot, and other fungi. When treated with ultraviolet irradiation it is converted into vitamin  $\text{D}_2$  [18, 19].



**Figure 3:** Structure of ergosterol (2).

## CONCLUSIONS

The edible mushrooms are known for their high content of steroidal compounds and unsaturated fatty acids. From a single fruit-body of the basidiomycete *Fistulina hepatica*, collected in the forestal region of Seitenbrueck, Germany were isolated for the first time a secondary metabolites which were separated and purified by preparative chromatographic methods. The structures of the both compounds were determined on the basis of NMR spectroscopic methods including 1D ( $^1\text{H}$ ,  $^{13}\text{C}$  and DEPT) and 2D (COSY, HSQC and HMBC) and by mass spectrometry data (ESI, EI and HREIMS). The new fatty acid metabolite, octadeca-8,11-dienoic acid methyl ester (**1**) was isolated together with the known ergosterol (**2**). The new unsaturated fatty acid is suggested to contribute much to the taste and flavour of basidiomycetes as food components and to pharmacological effects as well. The new fatty acid methyl ester (**1**) displays very low activity ( $\geq 1000 \mu\text{g/ml}$ ) against *Escherichia coli* and *Proteus vulgaris*. Ergosterol (**2**) does not occur in plant or animal cells. It is a component of yeast, mushrooms and other fungal cell membranes.

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